Preliminary communication

Syntheses of partially protected methyl α-D-mannopyranosides*

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In recent years, much emphasis has been placed on the synthesis of complex saccharides^{2,3}. The benzyl group has been used widely as a "persistent" protective group in various aglycon hydroxides suitable for such synthetic investigations in carbohydrate chemistry⁴. Recently, we reported selective monobenzylation of methyl 4-O-benzyl- α -L-rhamnopyranoside¹. Using methyl 4,6-di-O-benzyl- α -D-mannopyranoside (1) as a model compound, we have studied monobenzylation under the various conditions now described.

In one of the approaches, we employed phase-transfer catalysis, as introduced by Garegg et al.⁵, for selective monobenzylation of the 2-hydroxyl group. Thus, to a solution of compound 1 (2.6 g), tetrabutylammonium hydrogensulfate (0.48 g), and benzyl bromide (1.44 mL) in dichloromethane (120 mL) was added aqueous sodium hydroxide (10 mL of a 5% solution), and the mixture was boiled under reflux for 40 h. The usual processing gave a syrupy residue that was conveniently purified by chromatography on a column of silica gel, to afford product 2 having $[\alpha]_D^{25} + 14.4^{\circ}$ (c 1, CHCl₃); this was clearly distinguishable in t.l.c. from the known methyl 3,4,6-tri-O-benzyl- α -D-mannopyranoside⁶ (3) having $[\alpha]_D^{25} + 54.2^{\circ}$ (c 1, CHCl₃). An

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authentic sample of compound 2 was also obtained as follows. On heating with acetic acid, methyl 3-O-allyl-4,6-O-benzylidene- α -D-mannopyranoside ⁷ (7) gave methyl 3-O-allyl- α -D-mannopyranoside (8) as a syrup, $[\alpha]_D^{23} + 59.7^{\circ}$ (c 1, HCONMe₂), which, on benzylation in the presence of powdered potassium hydroxide in N,N-dimethylformamide, produced 9 in 49.5% yield; $[\alpha]_D^{24} + 35.3^{\circ}$ (c 1, CHCl₃). Removal of the 3-O-allyl group by treatment with tris(triphenylphosphine)rhodium chloride⁸ gave compound 2. The i.r. and p.m.r. spectra of the two samples prepared by the different methods were respectively superposable.

In another approach, a mixture of diol 1 (1 g), trityl chloride (0.92 g), triethylamine (0.75 mL), 4-(dimethylamino)pyridine⁹ (30 mg), and dichloromethane (50 mL) was refluxed under nitrogen for 20 h. The usual processing, and purification by column chromatography, provided pure compound 4 (0.6 g, 38%), which was recrystallized from acetone-pet. ether; m.p. 128-130°, [α]_D²³ +94.8° (c 1, CHCl₃). Compound 4 was characterized by conversion, in two steps, into the aforementioned methyl 2,4,6-tri-O-benzyl-α-D-mannopyranoside (2); benzylation of compound 4 under the usual conditions, followed by removal of the trityl group (with CF₃CO₂H-CHCl₃), afforded compound 2, thereby supporting the finding⁹ that tritylation of diol 1 occurs at the (equatorial) 3-hydroxyl group. The i.r. and ¹H-n.m.r. spectra of the compound obtained were identical to those of compound 2 prepared by the aforementioned routes.

For selective monobenzylation of the 3-hydroxyl group, diol 1 (1.07 g) was treated with bis(tributyltin) oxide (2.2 g) in toluene (30 mL) for 4 h at 140°, with continuous removal of water, to give, after evaporation of toluene, an intermediate, oily, stannylation product which was then heated with benzyl bromide (15 mL) under nitrogen for 30 h at 90°. The mixture was cooled, coevaporated several times with water, and finally with toluene, to give a syrup which, on column chromatography, afforded pure 3 (1.7 g, 73%); identical with an authentic sample of methyl 3,4,6-tri-O-benzyl-α-D-mannopyranoside⁶ on the basis of optical rotation, and i.r. and ¹H-n.m.r. spectroscopy.

The structures assigned to compounds 2 and 3 were confirmed by examination of the ¹H-n.m.r. spectra of their acetates (5 and 6, respectively). Acetylation (acetic anhydride-pyridine) of 2 gave syrupy 5, $[\alpha]_D^{24} + 9.1^{\circ}$ (c 1, CHCl₃); p.m.r. at 100 MHz (CDCl₃): τ 4.74 (dd, 1 H, $J_{3,4}$ 9 Hz, $J_{3,2}$ 3 Hz, H-3) and 8.1 (s, 3 H, equatorial OAc). Similar acetylation of 3 furnished syrupy 6, $[\alpha]_D^{24} + 29.8^{\circ}$ (c 1, CHCl₃); p.m.r. at 100 MHz (CDCl₃): τ 4.54 (dd, 1 H, $J_{2,3}$ 3 Hz, $J_{2,1}$ 2 Hz, H-2) and 7.84 (s, 3 H, axial OAc).

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